


<b>Name:</b>	 <b>UPES</b> UNIVERSITY OF TOMORROW
<b>Enrolment No:</b>	

**UNIVERSITY OF PETROLEUM AND ENERGY STUDIES**  
**End Semester Examination, May 2022**

**Course: Elements of Modern Physics**  
**Program: B.Sc (H) Physics**  
**Course Code: PHYS 2005**

**Semester : IV**  
**Time : 03 hrs.**  
**Max. Marks: 100**

**Instructions:**

- All questions are compulsory (Q.No. 9 and Q.No. 11 have an internal choice)
- Scientific calculators can be used for calculations

**SECTION A**  
**(5Q x 4M = 20 Marks)**

- All Questions are compulsory, Each Question carries 4 Marks
- Write very Short Answers/ Solve

Q. No.	Statement of question	Marks	CO
1	What is the photoelectric effect? What are the factors that influence the photoelectric effect?	4	CO1
2	Calculate the de-Broglie wavelength associated with a proton moving with a velocity equal to $(1/20)^{\text{th}}$ , of the velocity of light.	4	CO4
3	Uncertainty in the time of an excited atom is about $10^{-8}$ sec. What are the uncertainties in energy and frequency of radiation?	4	CO1
4	Explain pair production by gamma-ray photons in the vicinity of the nucleus.	4	CO4
5	Distinguish between spontaneous and stimulated emissions.	4	CO1

**SECTION B**  
**(4Q x 10M = 40 Marks)**

- All Questions are compulsory, Q.No. 9 has an internal choice, Each Question carries 10 Marks
- Write Short/ Brief notes/ Derive/ Solve

Q. No	Statement of Question	Marks	CO
6	What is Compton Effect? Explain the Compton effect with a neat diagram. Derive an equation for a shift in the Compton wavelength.	10	CO2
7	(a) Prove that the relation between the phase velocity ( $v_p$ ) and group velocity ( $v_g$ ) in a dispersive medium is $v_g = v_p - \lambda \left( \frac{dv_p}{d\lambda} \right)$ (b) Calculate the lowest energy of an electron confined in a 3-D cubical box of each side $1 \text{ \AA}$ .	10	CO2
8	(a) Mention any five properties of $\alpha$ -radiations.	10	CO2

	(b) The half-life of Radon is 3.8 days. After how many days will only one-twentieth of the radon sample be leftover. (answer upto the second decimal) (use $\log_{10} 20 = 1.3010$ ) (5)		
9	(a) Explain the construction and working of a pulsed laser with the help of a neat energy level diagram. <b>(OR)</b> (b) What are Einstein's Coefficients? Derive the relation between Einstein's coefficients.	10	CO3
<b>SECTION-C</b> <b>(2Q x 20M = 40 Marks)</b>			
<ul style="list-style-type: none"> <li>All Questions are compulsory, <b>Q.No. 11</b> has an internal choice, Each Question carries 20 Marks</li> <li>Write long answers/ Derive/ Solve</li> </ul>			
Q. No	Statement of question	Marks	CO
10	(a) Show that the wave function of a particle trapped into a one-dimension box of length $L$ is (10) $\Psi_n(x) = \sqrt{\frac{2}{L}} \sin\left(\frac{n\pi x}{L}\right), \text{ where } n=1, 2, 3, \dots$ (b) Electrons with energies of 1 eV and 2 eV are incident on a barrier 5 eV high and 5 Å wide. Find their respective transmission probabilities. (10)	20	CO3
11	(a) Define the range of the $\alpha$ -particle. Explain Gamow's theory of alpha decay with the necessary diagram. (10) (b) Explain the binding energy of the nucleus. Find the binding energy of an $\alpha$ -particle from the below-given data. Mass of Helium nucleus = 4.001265 a.m. u Mass of proton = 1.007277 a.m. u Mass of neutron = 1.008666 a.m. u 1 a.m. u = 931.4812 MeV (10) <b>(OR)</b> (a) What are various nuclear models? State and explain the liquid drop model of the nucleus, with the analogies between a small drop of a liquid and a nucleus. (10) (b) Derive a relation for the semi-empirical mass formula for the nucleus giving arguments for each of the terms involved. (10)	20	CO3

Values of Constants

Constant	Standard Values
Planck's Constant ( $h$ )	$6.63 \times 10^{-34}$ Joule – sec
Permittivity of free space ( $\epsilon_0$ )	$8.85 \times 10^{-12}$ Farad/meter
Velocity of light ( $c$ )	$3 \times 10^8$ m/sec
Boltzmann constant ( $k_B$ )	$1.38 \times 10^{-23}$ JK <sup>-1</sup>
Rest mass of an Electron	$9.11 \times 10^{-31}$ kg
Mass of the proton	$1.67 \times 10^{-27}$ kg
Charge of an electron	$1.6 \times 10^{-19}$ C