


| Name: | |  | |
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| Enrolment No: | | | |
| UPES End Semester Examination, December 2024 | | | |
| Programme Name: Integrated BSc-MSc (Physics) | | Semester: VII | |
| Course Name: Atomic & Molecular Physics | | Time: 3 hrs | |
| Course Code: PHYS4001 | | Max. Marks: 100 | |
| No. of pages: 2 | | | |
| Instructions: As instructed in each section. Symbols have their usual meanings. | | | |
| SECTION A (Answer all the questions: 5 Qs × 4 M = 20 Marks) | | | |
| S. No. | | Marks | CO |
| Q 1 | In a tabular form write down the differences between 3-level laser and 4-level laser. | 4 | CO1 |
| Q 2 | Identify which molecules are rotational active: (a) H ₂ , (b) CO ₂ , (c) H ₂ O, (d) CCl ₄ | 4 | CO1 |
| Q 3 | Sketch space quantization of J for the 4 ² P _{3/2} state. | 4 | CO2 |
| Q 4 | A vibrational level is observed at 2900 cm ⁻¹ for HCl molecule. Estimate the force constant k (consider, atomic mass unit = 1.677×10 ⁻²⁷ kg). | 4 | CO3 |
| Q 5 | Compute free space electric and magnetic fields associated with a He-Ne laser of intensity 15 W/m ² . | 4 | CO4 |
| SECTION B (Answer all the questions: 4 Qs × 10 M = 40 Marks) | | | |
| Q 6 | What is population inversion? Discuss details of how population is achieved in 3-level and 4-level laser with the help of suitable diagrams. | 10 | CO1 |
| Q 7 | Describe construction and working of solid state laser (ruby laser). | 10 | CO2 |
| Q 8 | Calculate observed line frequencies for the atoms placed in 5.25 Tesla magnetic field. The fundamental line frequency is 600×10 ¹² Hz. | 10 | CO2 |
| Q 9 | Discuss Paschen-Back effect. OR Discuss anomalous Zeeman effect. | 10 | CO3 |

SECTION C

(Answer all the questions: 2 Qs × 20 M = 40 Marks)

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| Q 10 | (a) Show that Einstein co-efficients A and B are related to each other in the following form, $A_{21} = \frac{8\pi h\nu^3}{c^3} B_{21}, \text{ and } B_{12} = B_{21}$ | 10 | CO3 |
| | (b) A rotational structure of diatomic molecule shows microwave absorption with rotational constant $B = 12 \text{ cm}^{-1}$. Compute moment of inertia (I) of the diatomic molecule. | 10 | CO4 |
| Q 11 | (a) Deduce the expression for magnetic moment (μ) originating due to orbital motion of electron. | 10 | CO3 |
| | (b) Analyse, P, R and Q branches of vibrational-rotational spectra of diatomic molecule. | 10 | CO4 |
| | OR | | |
| | (a) Write short note on E.S.R. | 10 | CO3 |
| | (b) Calculate 'g' and 'gm _j ' and magnetic moment for the state, $3^2P_{3/2}$. | 10 | CO4 |